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To answer the question, you need to know the electron configuration of the atoms to see which one has 7 unbonded electrons, like chlorine. Fluorine, element number 9, has 2 electrons in the s sublevel (K shell). The L shell is incompletely filled, with 7 electrons. 9.

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38. The ground state electron configuration of Fe is \_\_\_\_\_. A. 2 1s 2s 2 3s 3p 3d6 B. 2 1s 2 2s 2p6 3s2 3p6 4s 3d6 C. 2 1s2 2s 2p6 3s2 3p6 4s2 D. 2 1s 2 2s 2p6 3s 3p6 4s2 4d6 3p. The ground state electron configuration of Ga is \_\_\_\_\_. A. 2 1s 2s2 3s2 3p6 3d10 4s2 4p1 B. 6 1s 2 2s 2p 3s2 3p6 4s2 4d10 4p1 C. 2 1s2 2s 2p6 3s2 3p6 4s2 3d10 4p1

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Which of the following is the correct electron configuration for the bromide ion, Br-? a) [Ar] 4s 2 4p 5. b) [Ar] 4s 2 3d 10 4p 5. c) [Ar] 4s 2 3d 10 4p 6. d) [Ar] 4s 2 3d 10 4p 6 5s 1. e) [Ar] 4s 2 3d 10 3p 6. answer . text explanation. 6-89 . Which of the following elements has the largest number of electrons for which the principal quantum ...